

LAMPIRAN

Lampiran 1. Perhitungan Pembuatan Larutan

a) Pembuatan larutan 0,5 M $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$

$$M = \frac{\text{massa}}{Mr} \times \frac{1000}{v}$$

$$0,5 \text{ M} = \frac{\text{massa}}{270,32 \text{ g/mol}} \times \frac{1000 \text{ mL}}{50 \text{ mL}}$$

$$\text{massa} = \frac{270,32 \text{ g/mol} \times 50 \text{ mL}}{1000 \text{ mL}} \times 0,5 \text{ M}$$

$$\text{massa} = 6,758 \text{ g}$$

b) Pembuatan larutan 0,25 M $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$

$$M = \frac{\text{massa}}{Mr} \times \frac{1000}{v}$$

$$0,25 \text{ M} = \frac{\text{massa}}{278,06 \text{ g/mol}} \times \frac{1000 \text{ mL}}{50 \text{ mL}}$$

$$\text{massa} = \frac{278,06 \text{ g/mol} \times 50 \text{ mL}}{1000 \text{ mL}} \times 0,25 \text{ M}$$

$$\text{massa} = 3,476 \text{ g}$$

c) Pembuatan larutan 0,5 M NaOH

$$M = \frac{\text{massa}}{Mr} \times \frac{1000}{v}$$

$$0,5 \text{ M} = \frac{\text{massa}}{40 \text{ g/mol}} \times \frac{1000 \text{ mL}}{250 \text{ mL}}$$

$$\text{massa} = \frac{40 \text{ g/mol} \times 250 \text{ mL}}{1000 \text{ mL}} \times 0,5 \text{ M}$$

$$\text{massa} = 5 \text{ g}$$

Lampiran 2. Persamaan Scherrer

$$D = \frac{K \lambda}{\beta \cos \theta}$$

Keterangan:

D = Ukuran kristal

λ = Panjang gelombang (1,54056 Å)

K = Shape factor (0,9)

β = FWHM (rad)

θ = Sudut bragg (°)

a) Fe_3O_4 -NPs

$$D1 = \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ Å}}{0,008901179 \cdot 0,965608898} = 161,3140977 \text{ Å}$$

$$D2 = \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ Å}}{0,008726646 \cdot 0,952156066} = 166,8651389 \text{ Å}$$

$$\begin{aligned}
 D3 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,005759587 \cdot 0,929326043} = 259,0369476 \text{ \AA} \\
 D4 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,007853982 \cdot 0,877648572} = 201,1456278 \text{ \AA} \\
 D5 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,012042772 \cdot 0,853959734} = 134,820917 \text{ \AA}
 \end{aligned}$$

b) Fe₃O₄-NPs/CDs 15 mL

$$\begin{aligned}
 D1 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,00837758 \cdot 0,965540798} = 171,408318 \text{ \AA} \\
 D2 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,008726646 \cdot 0,952102712} = 166,8744897 \text{ \AA} \\
 D3 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,026179939 \cdot 0,928906533} = 57,01386495 \text{ \AA} \\
 D4 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,010297443 \cdot 0,877355617} = 153,4673825 \text{ \AA} \\
 D5 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,007330383 \cdot 0,853459851} = 221,6212351 \text{ \AA}
 \end{aligned}$$

c) Fe₃O₄-NPs/CDs 30 mL

$$\begin{aligned}
 D1 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,009948377 \cdot 0,965267733} = 144,3846805 \text{ \AA} \\
 D2 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,009075712 \cdot 0,952102712} = 160,456241 \text{ \AA} \\
 D3 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,009075712 \cdot 0,929100302} = 164,4287727 \text{ \AA} \\
 D4 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,010297443 \cdot 0,877564904} = 153,4307828 \text{ \AA} \\
 D5 &= \frac{K \lambda}{\beta \cos \theta} = \frac{0,9 \cdot 1,54056 \text{ \AA}}{0,009599311 \cdot 0,853596261} = 169,2109895 \text{ \AA}
 \end{aligned}$$

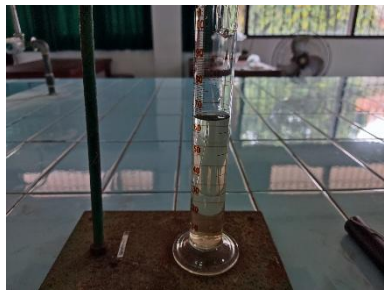
Lampiran 3. Dokumentasi Penelitian



Alat hydrothermal autoclave



Sintesis CDs metode hydrothermal



Menyaring CDs hasil sintesis



CDs di bawah sinar UV



Larutan prekursor

*Black liquid*

Pencampuran prekursor

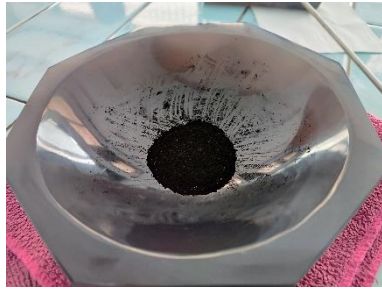
Setelah penambahan *black liquid*

Setelah penambahan NaOH



Setelah pH mencapai 11

Koloid Fe_3O_4 -NPs Fe_3O_4 -NPs setelah dioven



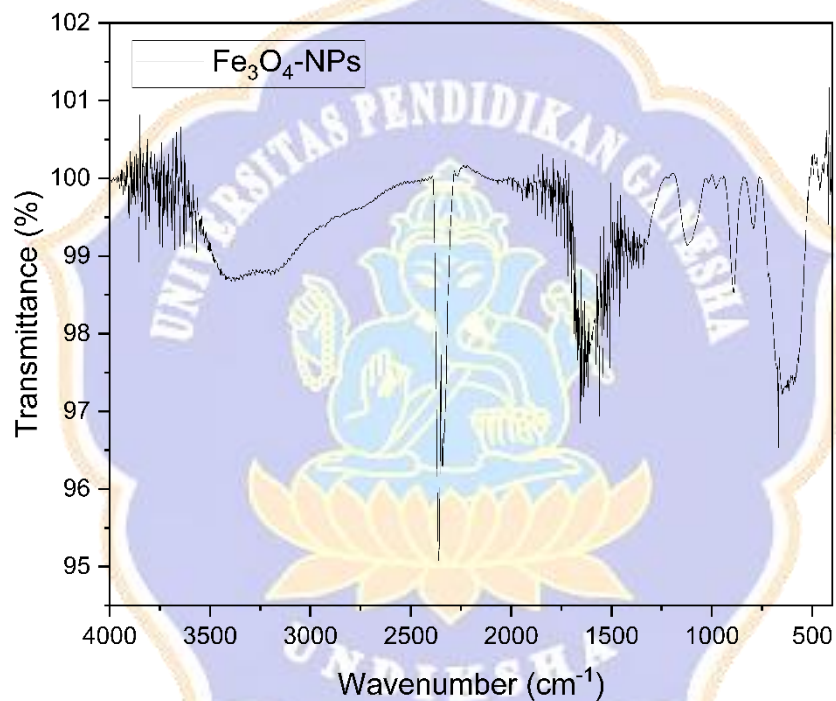
Penggerusan Fe_3O_4 -NPs

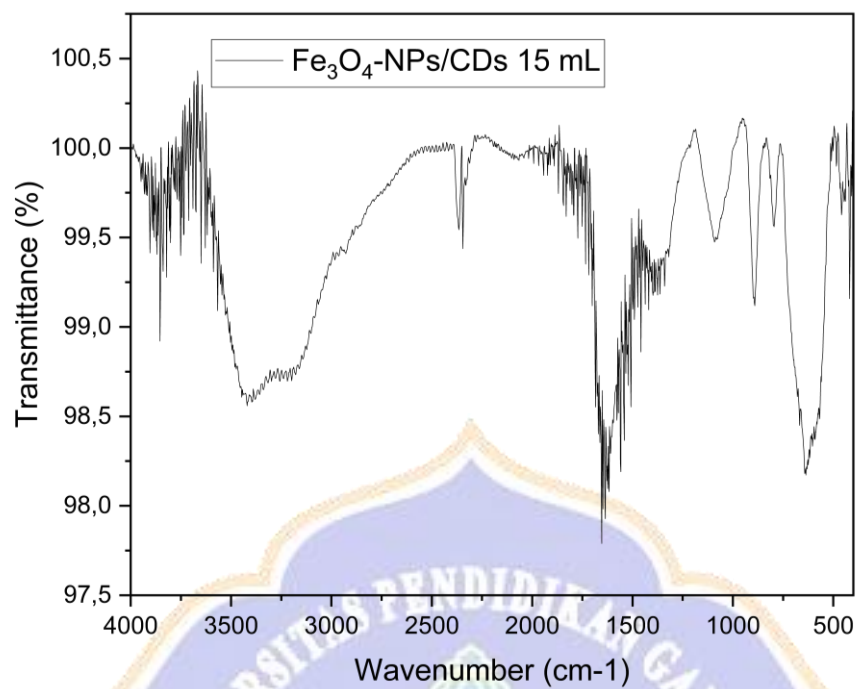
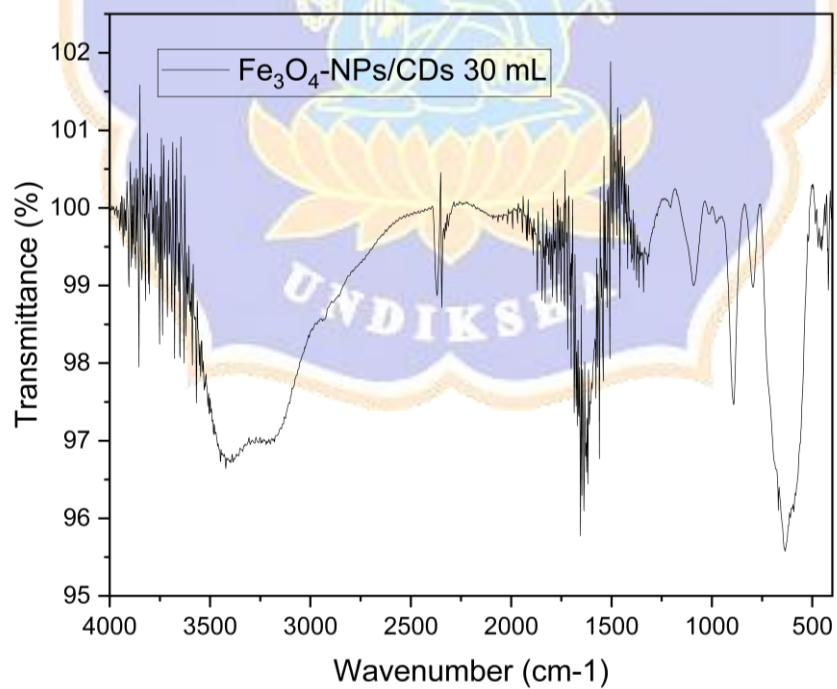


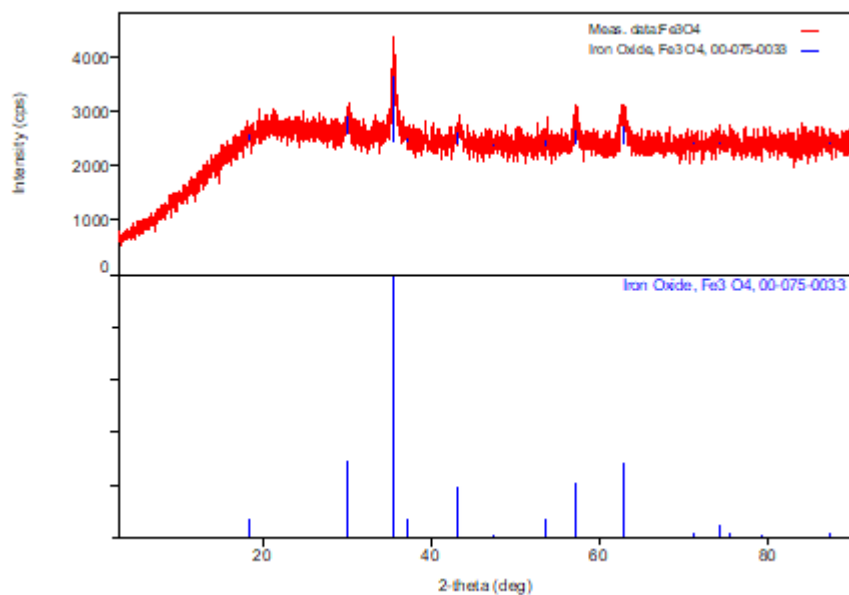
Fabrikasi Fe_3O_4 -NPs dengan CDs dengan sonikasi

Lampiran 4. Karakterisasi Nanopartikel

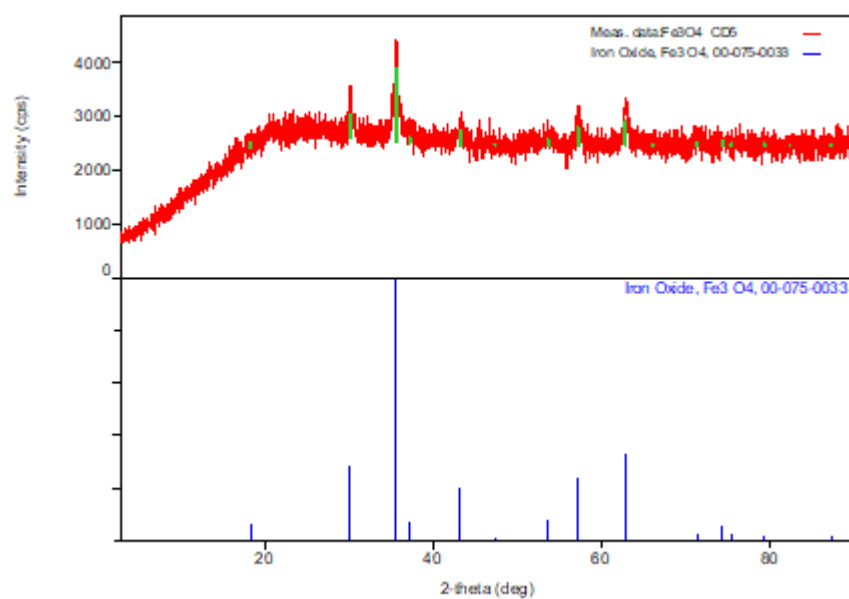
FTIR Fe_3O_4 -NPs



FTIR Fe_3O_4 -NPs/CDs 15 mLFTIR Fe_3O_4 -NPs/CDs 30 mL

XRD Fe_3O_4 -NPs

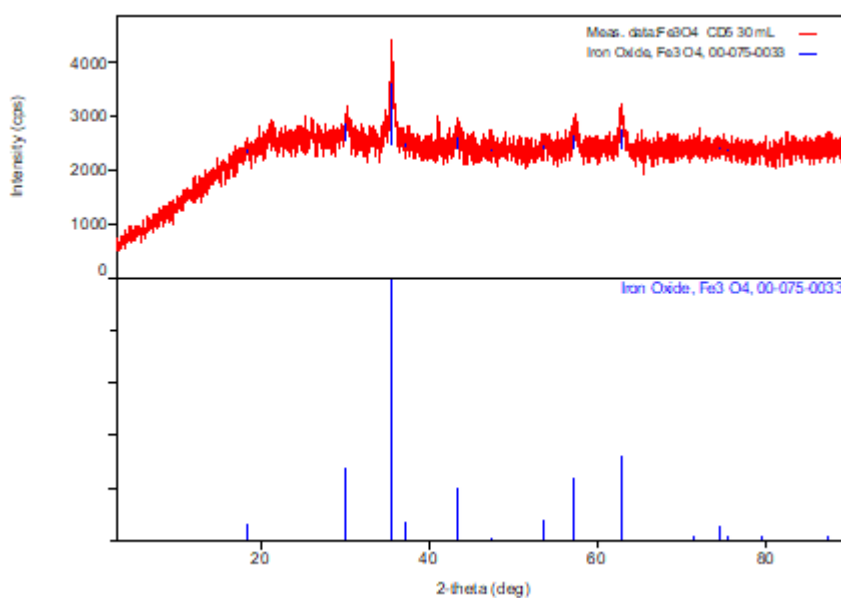
No.	2-theta(deg)	d(ang.)	Height(cps)	FWHM(deg)	Int. I(cps deg)	Int. W(deg)	Asym. factor
1	30.14(2)	2.962(2)	300(50)	0.51(7)	163(29)	0.55(19)	0.6(4)
2	35.59(3)	2.520(2)	1143(98)	0.50(4)	974(28)	0.85(10)	1.0(3)
3	43.34(3)	2.0863(14)	266(47)	0.33(8)	176(22)	0.7(2)	2(2)
4	57.28(4)	1.6071(11)	377(56)	0.45(7)	252(27)	0.67(17)	2.9(14)
5	62.71(2)	1.4804(4)	433(60)	0.69(7)	391(30)	0.90(19)	0.6(4)

XRD Fe_3O_4 -NPs/CDs 15 mL

No.	2-theta(deg)	d(ang.)	Height(cps)	FWHM(deg)	Int. I(cps deg)	Int. W(deg)	Asym. factor
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1	30.17(2)	2.960(2)	433(60)	0.48(10)	374(29)	0.86(19)	0.5(4)
2	35.61(3)	2.5190(19)	1184(99)	0.50(5)	1122(34)	0.95(11)	0.9(2)
3	43.47(6)	2.080(3)	204(41)	1.5(3)	643(56)	3.2(9)	5(4)
4	57.35(4)	1.6052(11)	371(56)	0.59(8)	324(31)	0.9(2)	2.9(10)
5	62.82(4)	1.4781(9)	601(71)	0.42(10)	463(27)	0.77(14)	0.6(4)

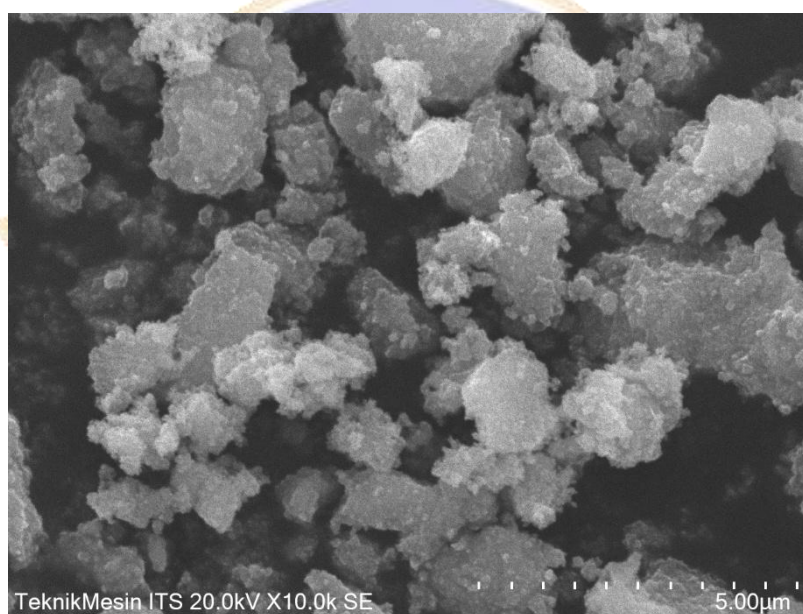
XRD Fe₃O₄-NPs/CDs 30 mL



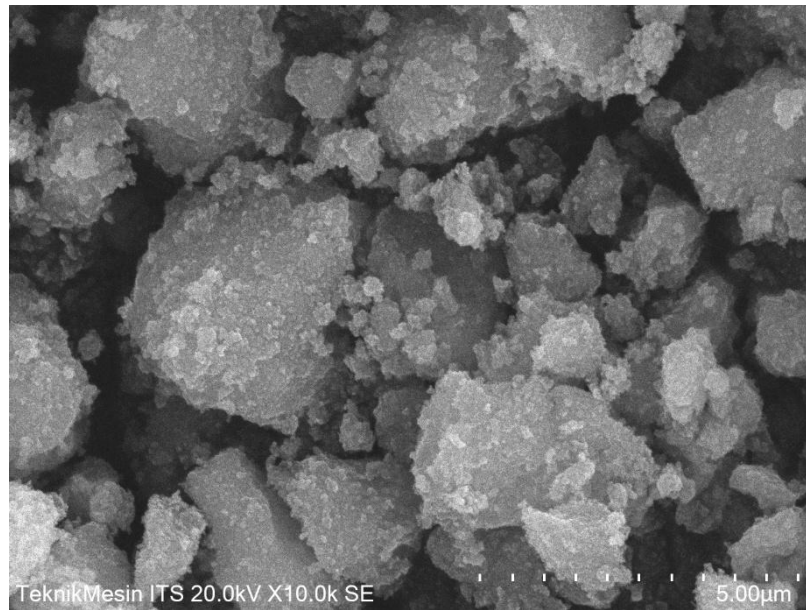
No.	2-theta(deg)	d(ang.)	Height(cps)	FWHM(deg)	Int. I(cps deg)	Int. W(deg)	Asym. factor
1	30.29(10)	2.948(9)	289(49)	0.57(11)	236(40)	0.8(3)	1.2(10)
2	35.61(3)	2.519(2)	1066(94)	0.52(4)	944(36)	0.89(11)	0.8(3)
3	43.41(4)	2.0830(19)	234(44)	0.52(19)	227(38)	1.0(3)	2(4)
4	57.33(11)	1.606(3)	273(48)	0.59(9)	194(35)	0.7(3)	1.3(10)
5	62.79(7)	1.4788(15)	444(61)	0.55(10)	328(37)	0.74(18)	0.5(4)

SEM-EDS Fe₃O₄-NPs**Smart Quant Results**

Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	A	F
C K	17.53	33.80	104.92	9.56	0.0536	1.1535	0.2650	1.0000
O K	29.29	42.42	407.23	8.17	0.1061	1.1078	0.3271	1.0000
NaK	1.62	1.63	12.31	15.42	0.0029	1.0099	0.1764	1.0005
AlK	0.87	0.75	18.47	13.04	0.0035	0.9910	0.4028	1.0019
SiK	0.51	0.42	14.76	14.03	0.0027	1.0136	0.5327	1.0032
ClK	0.58	0.38	18.05	12.33	0.0047	0.9463	0.8361	1.0134
CrK	1.03	0.46	20.26	9.21	0.0103	0.8639	1.0077	1.1499
FeK	48.57	20.15	656.22	1.95	0.4247	0.8597	1.0109	1.0062

SEM-EDS Fe₃O₄-NPs/CDs 15 mL**Smart Quant Results**

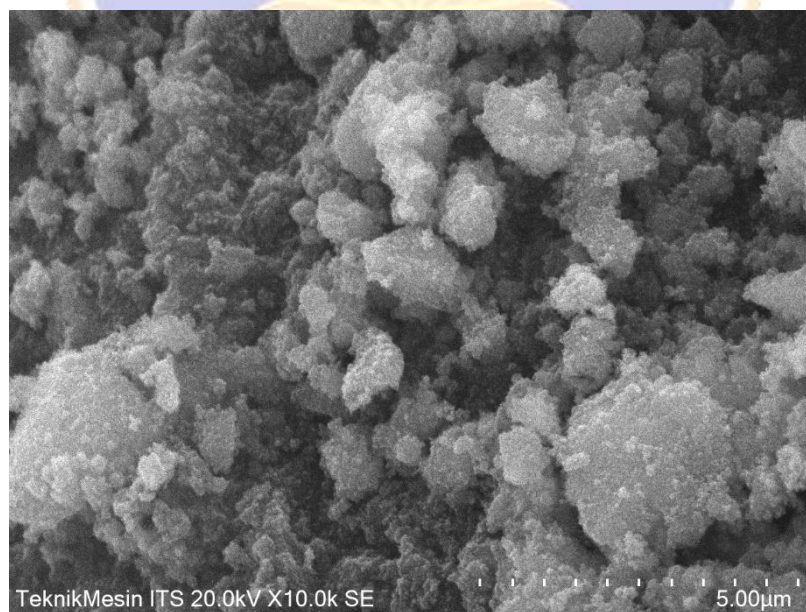
Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	A	F
C K	16.56	32.38	114.32	9.34	0.0542	1.1568	0.2830	1.0000
O K	31.04	45.58	492.39	7.94	0.1192	1.1111	0.3455	1.0000
FeK	52.40	22.04	766.83	1.93	0.4608	0.8628	1.0137	1.0054



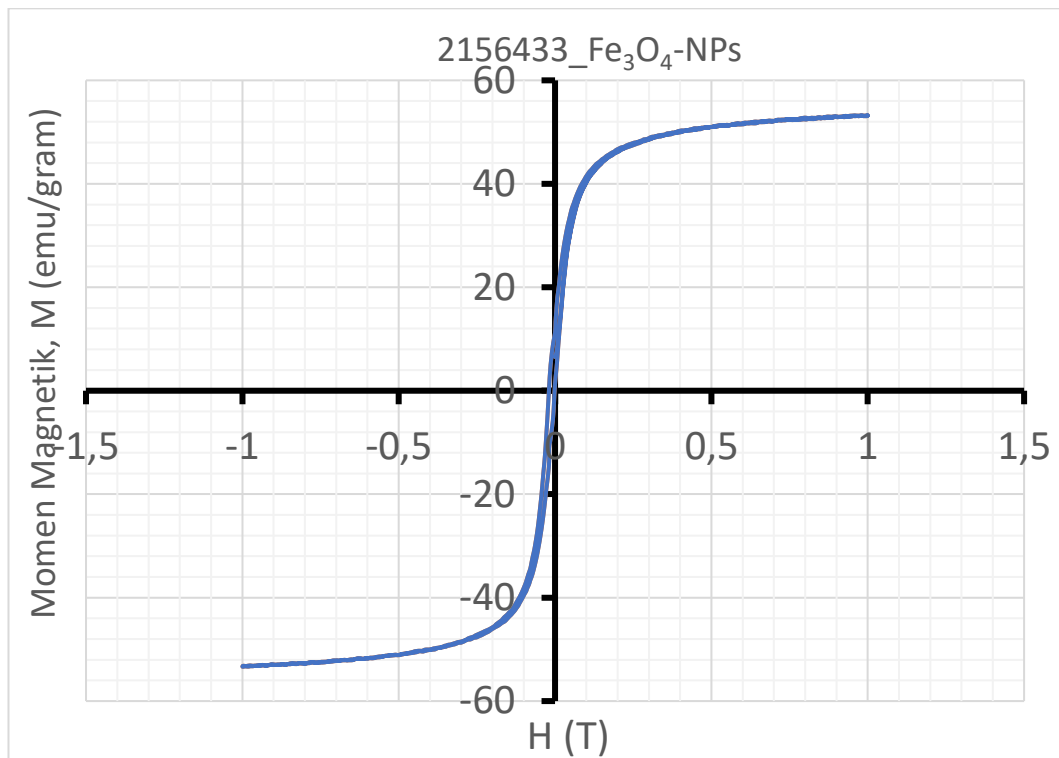
SEM-EDS Fe₃O₄-NPs/CDs 30 mL

Smart Quant Results

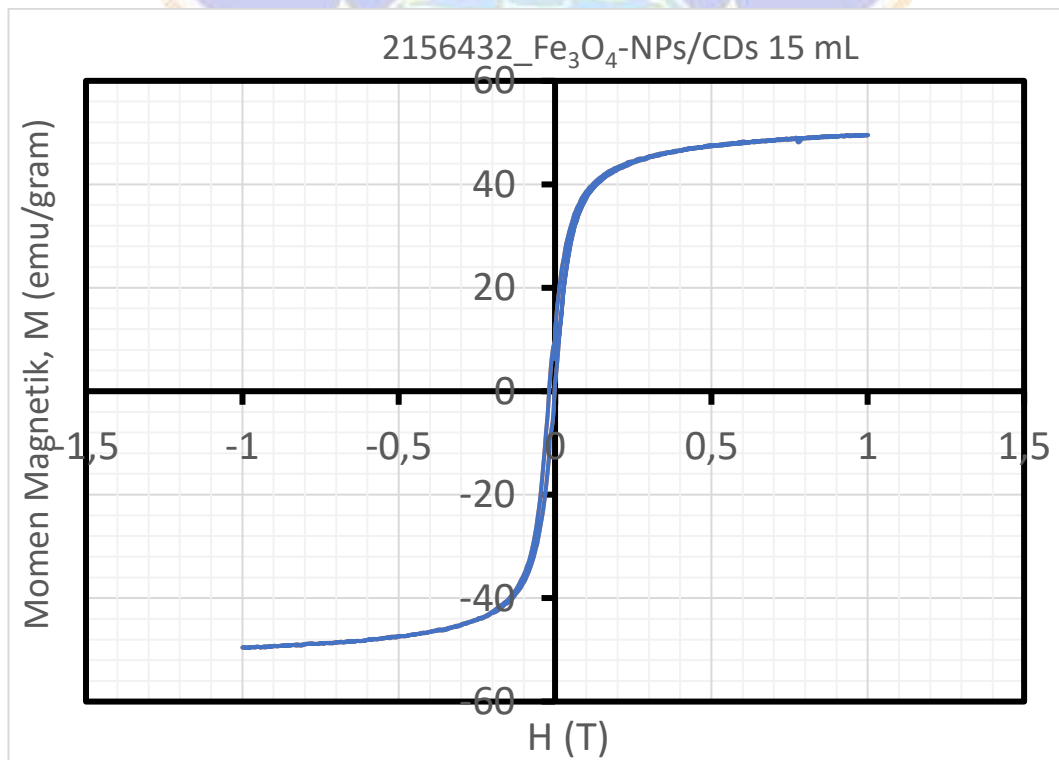
Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	A	F
C K	14.47	30.26	74.46	9.56	0.0460	1.1723	0.2711	1.0000
O K	27.93	43.84	356.27	7.84	0.1123	1.1263	0.3570	1.0000
FeK	57.59	25.90	655.28	1.89	0.5130	0.8761	1.0122	1.0044



VSM Fe_3O_4 -NPs



VSM Fe_3O_4 -NPs/CDs 15 mL



VSM Fe_3O_4 -NPs/CDs 30 mL

